Obtaining Ising-like expansions for binary alloys from first principles

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Many measurable properties of crystalline binary  $A_1 X B_x B_x$  alloys, such as phase diagrams and excess thermodynamic functions, could be predicted via lattice statistical mechanics methods if one knew the 'configurational energy'. The latter describes the energy at  $T = 0$  for each of the  $2^N$  possible occupation patterns of the *N* lattice sites by an *A* or a *B* atom. Traditional approaches described the configurational energy either via empirically fitted, truncated Ising Hamiltonians, or through highly approximated coherent-potential constructs. We illustrate here the alternative approach of 'mixed-basis cluster expansion' which extracts from a set of *ab initio* local density approximation calculations of

The configurational energy  $E_{\text{config}}[S_i, R_i]$  can be modelled at different levels of approximation. The distinguishes between 'direct approaches' and 'cluster expansions' (CE).

of the system. Finally, such simulations give directly the short-range-order parameters [8, 9]  $(x, T)$  and mixing enthalpies [10, 11].

and the 'concentration wave method' [22], which involve various approximations beyond the first-principles local density approximation (LDA) that underlies these approaches. In their various forms, these methods ignored atomic relaxation of even size-mismatched systems, approximated the total energy just by the sum of energy eigenvalues, and until recently [23], overlooked the Madelung contribution to the energy of ionic alloys. Here we will discuss how a robust CE can be obtained directly from LDA calculations on a few ordered *ApBq* structures. This 'mixed-basis cluster-expansion' (MBCE) approach [1, 24] builds on and extends the Connolly–Williams [25] approach. We have recently applied this method to predict thermodynamic behaviour of numerous binary alloys, including Cu–Au [26–28], Cu–Ag [26], Cu–Pt [29, 30], Ni–Au [26, 28, 31], Ag–Au [32, 33], Cu–Pd [33], Ni–V [34, 35], Ni–Pt [36], Ag–Pd [37, 38], Al–Zn [10, 39], Pd–V [34, 35], Pd–Pt [32], Cu–Al [10] and Cu–Zn [40] and some semiconductor alloys [20, 41, 42]. Such CEs are then used in Monte-Carlo simulations of the Hamiltonian. This yields phase diagrams, ground state structures, thermodynamic functions, short-range-order profiles and precipitate shapes.

In this paper, we illustrate in detail how such an expansion is constructed from LDA total energies. We focus on the technical issues of how a robust fit is achieved, how structures are chosen and how a stable expansion is obtained. We illustrate this using three systems: Ni–Pt, Cu–Au and Sc<sub>1</sub>  $\boldsymbol{x}$   $\boldsymbol{x}$  S (where denotes a vacancy on the Sc site)

pure *A* and *B* cannot be expressed by these finite-ranged *J* s. Consequently, this contribution, *E*<sub>CS</sub>( $θ$ ), to the formation enthalpy is expressed by the last term in equation (2):

$$
E_{\text{CS}}( \ ) \quad \sum_{k} \frac{E_{\text{CS}}^{\text{eq}}(x,\bar{k})}{4x(1-x)} S(k, \quad )^2 F(k), \tag{5}
$$

where  $E_{CS}^{eq}(x, \bar{k})$  is the *constituent strain energy* [8,24], which is defined as the strain energy required to maintain coherency along an interface (with orientation  $\bar{k}$ ) of bulk *A* and *B*.

To understand the necessity for this term in the CE, consider coherent phase separation, i.e. solid *A* and *B* coherently match along the crystallographic direction  $\bar{k}$  (formally, this is a long-period superlattice  $A_n/B_n$  with  $n \to \infty$  oriented along  $\bar{k}$ 

## **3. Determination of the expansion coefficients in the MBCE**

## *3.1. The constituent strain*

Determining  $E_{CS}^{eq}(x, \bar{k})$  requires three steps: (i) calculating the epitaxial energies of each end point, *A* and *B* for several directions  $\bar{k}$  and in-plane lattice constant *a*, (ii) finding the *a* that minimizes the total  $A + B$  epitaxial energies for intermediate concentrations and (iii) interpolating the results of (ii) to arbitrary directions of  $\bar{k}$ .

For the first step, one calculates via LDA for pure *A* and pure *B* the total energy of several different in-plane lattice constants,  $a$  (perpendicular to  $\bar{k}$ ), and in each case, the unit cell is

five principal directions. Naturally, each of the energies  $E_A^{\text{epi}}$  and  $E_B^{\text{epi}}$  is positive definite and, hence, the coherency strain of equation (7) must be positive definite.

Finally, in the third step, the constituent strain energy is interpolated to arbitrary directions of  $\bar{k}$  by fitting the results to an expansion of Kubic harmonics. That is, after  $E_{CS}^{eq}(x, \bar{k})$  has been directly calculated for a set of directions using total energy methods, it is then interpolated to all directions by fitting the directly calculated results the following expression:

$$
E_{\rm CS}^{\rm eq}(x,\,\bar{k}) \, \sum_{l=0}^{l_{\rm max}} b_l(a \, ) K_l(\bar{k}). \tag{8}
$$

In cubic alloys, only terms with  $l = 0, 4, 6, 8, 10, 12, \ldots$  are non-zero and only these enter into the expansion. If only the first two terms are retained, then (8) reduces to well-known expression from harmonic elasticity theory (see discussion in [44]). However, our experience shows that anharmonic effects are usually significant, and so, in practice, we normally fit the 8. Figure 4 depicts  $E_{CS}^{eq}(x, \bar{k})$  for data using at least four terms in equation (8), i.e.  $I_{\text{max}}$ Ni-Pt and Cu-Au as a parametric plot in all directions.

In summary,  $E_{CS}^{eq}(x, \bar{k})$  is determined as follows:

(a) Epitaxial calculations are performed for each constituent of the alloy,  $A$  and  $B$ . For a series of different in-plane lattice constants  $a$  ( $a<sub>A</sub>$  a  $a_B$ ), the total energy is minimized by varying the out-of-plane lattice constant (parallel to  $\bar{k}$ ) while the in-plane lattice constant

 $\bigvee$ 

is held fixed. These values,  $E_{epi}$  (*a*), are interpolated to all values between  $a_A$  and  $a_B$  by a polynomial fit.

- (b) The strain energy for any composition *x* is taken to be the weighted average of the epitaxial energies calculated in (a) but we choose the in-plane lattice constant *a* to minimize the strain energy as shown in equation (7).
- (c) The strain energy for several directions  $\bar{k}$  is determined by repeating (a) and (b) for each direction  $\bar{k}$ . In practice, we use 5–7 different  $\bar{k}$ -directions. These directly calculated  $k$ -dependent values for the strain energy are then interpolated to arbitrary values of  $k$ by fitting to an expansion in Kubic harmonics, as shown in equation (8). Because of anharmonic effects, we typically find it necessary to use at least four terms in equation (8).

## *3.2. The constrained CE fit*

In constructing the fit of  $H_{CE}$  (*f*) to  $H_{LDA}$  (*f*), it is advantageous to use a different number interaction energies *J* than the number of LDA-calculated input formation enthalpies. This is unlike the Connolly–Williams approach [25] where the number of input structures and interaction energies must be the same, so a large number of LDA calculations might be needed to obtain a converged expression. However, at the same time, we must avoid 'over-fitting' by using too many interaction energies which results in a very accurate *fit* but a very poor *prediction* for structures not included in the fit. To achieve these objectives, we can require that  $J_{\text{pair}}(k)$  be a smooth function of  $k$ . We define [24] a 'smoothness value'  $M$  as

$$
M = \frac{1}{k} \sum_{k} J(k) \left[ -\frac{2}{k} \right]^{2} J(k), \tag{9}
$$

where the exponent

[40] and table III in [28] give a typical list of input structures. These structures need not be

## *4.2. Selection of the type of cluster interactions*

The interaction energies  $J$  are determined by minimizing equation (10).  $\bullet$  course an excellent fit can be obtained by using a large number of fitting parameters but such 'over-fitting' destroys the predictive accuracy of the CE. We desire a CE which accurately fits the input structures *and*







**Figure 11.** Ground state search for Cu–Au (see caption of figure 10).



**Figure 12.** Ground state search for Sc<sub>1</sub> *x*  $\Box$ *x*S (see caption of figure 10).

**Table 2.**



